

**5.61 Fall 2017  
Problem Set #3**

1.
  - A. McQuarrie, page 120, #3-3.
  - B. McQuarrie, page 120, #3-4.
  - C. McQuarrie, page 182, #4-11.
2. McQuarrie, pages 121-122, #3-11.
3.
  - A. McQuarrie, page 123, #3-17.
  - B. McQuarrie, page 127, #3-36.
4.
  - A. McQuarrie, page 122, #3-12. Answer this problem qualitatively by drawing a cartoon for  $n = 2$  and  $n = 3$  states.
  - B. Is there a simple mathematical/physical reason why the probabilities are not  $1/4$  for all four regions:  $0 \leq x \leq a/4$ ,  $a/4 \leq x \leq a/2$ ,  $a/2 \leq x \leq 3a/4$ , and  $3a/4 \leq x \leq a$ ?  
[HINT: where are the nodes in  $\psi_n(x)$ ?]
5. Solve for the energy levels of the particle confined to a ring as a crude model for the electronic structure of benzene. The two dimensional Schrödinger Equation, in polar coordinates, is

$$-\frac{\hbar^2}{2\mu} \left[ \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial}{\partial r} \right) + \frac{1}{r^2} \frac{\partial^2}{\partial \phi^2} + U(r, \phi) \right] \psi = E\psi.$$

For this problem,  $U(r, \phi) = \infty$  for  $r \neq a$ , but when  $r = a$ ,  $U(a, \phi) = 0$ .

- A. This implies that  $\psi(r, \phi) = 0$  for  $r \neq a$ . Why?
- B. If  $\psi(r, \phi) = 0$  for  $r \neq a$ , then  $\frac{\partial \psi}{\partial r} = 0$ . What is the simplified form of the Schrödinger Equation that applies when the particle is confined to the ring?
- C. Apply the “periodic” boundary condition that  $\psi(a, \phi) = \psi(a, \phi + 2\pi)$  to obtain the  $E_n$  energy levels.
- D. The C–C bond length in benzene is  $1.397 \text{ \AA}$ . Thus a circle which goes through all 6 carbon atoms has a radius  $r = 1.397 \text{ \AA}$ . Use this to estimate the  $n = 2 \leftarrow n = 1$  electronic transition for “benzene” treated as an electron on a ring. The longest wavelength allowed electronic transition for real benzene is at  $2626 \text{ \AA}$ . Explain why the agreement is not perfect.

6. 1-Dimensional Infinite Wells with Steps

Consider the potential

$$\begin{aligned} V(x) &= \infty & x < 0, x > a \\ V(x) &= 0 & 0 \leq x \leq a/2 \\ V(x) &= V_0 = \frac{\hbar^2}{8ma^2} (2)^2 & a/2 < x \leq a \end{aligned} .$$

- A. Sketch  $V(x)$  vs.  $x$ .
- B. What are the boundary conditions for  $\psi(x)$  at  $x = 0$  and  $x = a$ ?
- C. What requirements must be satisfied at  $x = a/2$ ?
- D. Solve for the  $n = 2$  (one node) and  $n = 3$  (two nodes)  $\psi_n(x)$  eigenfunctions of  $\hat{H}$  and  $E_n$  energy levels.

Hints: (i) For  $0 \leq x \leq a/2$ ,  $\psi_I(x) = A \sin k_I x$   
 $k_I = [2mE/\hbar^2]^{1/2}$

(ii) For  $a/2 < x \leq a$ ,  $\psi_{II}(x) = B \sin k_{II}(a - x)$   
 $k_{II} = [2m(E - V_0)/\hbar^2]^{1/2}$

(iii)  $\psi_I(a/2) = A \sin(k_I a/2)$   
 $\psi_{II}(a/2) = B \sin(k_{II} a/2)$   
 $\left. \frac{d\psi_I}{dx} \right|_{x=a/2} = Ak_I \cos(k_I a/2)$   
 $\left. \frac{d\psi_{II}}{dx} \right|_{x=a/2} = -Bk_{II} \cos(k_{II} a/2)$

- E. Compare your values of  $E_2$  and  $E_3$  to what you obtain from the de Broglie quantization condition

$$(n/2) = \frac{a/2}{\lambda_{n,I}} + \frac{a/2}{\lambda_{n,II}}$$

$$\lambda = h/p = 2\pi/k = h[2m(E - V(x))]^{-1/2}$$

- F. For the  $n = 2$  and  $n = 3$  energy levels, what are the probabilities,  $P_2$  and  $P_3$ , of finding the particle in the  $0 \leq x \leq a/2$  region?

- G.** (optional) Will the  $n = 2$  and  $3$  energy levels of the  $V_1(x)$  and  $V_2(x)$  potentials (defined below) be identical, as suggested by part **E**? Why?

$$V_1(x): \quad V_1(x) = \infty \quad x < 0, x > a$$

$$V_1(x) = 0 \quad 0 \leq x \leq a/2$$

$$V_1(x) = V_0 \quad a/2 < x \leq a$$

versus

$$V_2(x): \quad V_2(x) = \infty \quad x < 0, x > a$$

$$V_2(x) = 0 \quad 0 \leq x \leq a/4, 3a/4 \leq x \leq a$$

$$V_2(x) = V_0 \quad a/4 < x < 3a/4$$

- H.** Solve for the  $n = 1$   $\psi_1(x)$  and  $E_1$  for  $V_1$ .

**HINTS:** For  $a/2 < x \leq a$ ,  $\psi_{II}(x) = Be^{\kappa_{II}(a-x)} + Ce^{-\kappa_{II}(a-x)}$

$$\kappa_{II} = [2m(V_0 - E)/\hbar^2]^{1/2}$$

- I.** (optional) Is  $E_1$  for  $V_1$  larger or smaller than  $E_1$  for  $V_2$ ? Why? A cartoon will be helpful.

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